

# Prode Properties Test File

## for Prode Properties ver. 1.2b

This file may be used to insure that the Mathcad prode.dll file is in the right directory and that the functions are working properly. Prode Properties may be tested independent of Mathcad using the Excel version in the Prode directory. Examples are provided to demonstrate how the functions may be used. The process examples (compressor, nozzles, etc.) were based on the Excel examples included in the Prode Properties installation.

### Directions for using this worksheet

The default archive from Prode, def.ppp, will be used as the starting file. This worksheet will write changes to a new archive, test.ppp. This file will be placed in the same directory as def.ppp, but the user may also later archive to other directories.

Operations that set a variable will show a result of 1 if successful or 0 if not. Results that retrieve values will show the retrieved value, or 0 if no value is available. The exceptions to this convention will be noted.

Automatic calculation has been turned off so the new user may read these instructions before starting the computations.

### **Procedure:**

1. Calculate the entire worksheet, ctrl-F9
2. Select dep.ppp archive when first window appears, and click Open
3. When second popup window, the Prode archive, appears, it should show stream 2. Click OK.
4. Scroll through the worksheet to check for errors. See "Errors" below.
5. If the Prode window pops up, it is allowing you to view a recently created stream. Click OK to close.

### **Errors:**

- Mathcad will show errors in red as usual. Typical errors might be caused by syntax in the argument list, or the function may not be found in the library delivered by Prode, ppp.lib.
- If the result is zero, then the Prode function had an error or could not return a value. Frequently, this may be caused by the lack of a particular phase needed for an operation. For example, solid properties can't be returned when a stream has no solid phase, or when the temperature is above the melting point for a pure compound property.
- Rarely, a ppp.dll error window may appear with "Error accessing component's data archive". This appears to be caused by a lack of data in the chem.dat file for that particular property. If this window appears, it must be closed to proceed with the computations. See also the mc\_defErrMsg function for a way to prevent these windows from stopping the calculations. The program is set to prevent these error windows.

## File open commands

`mc_AFOpen("C:\ProgramData\prode\def.ppp") = 1` ■ This command sets the path to the archive file and directory. The path shown is the default set during the Prode Properties installation. This command will affect all subsequent uses of Prode on this computer until the path is changed again by another AFOpen command. **Therefore, this command should be used with caution.**

disabled

`mc_AOpen("dummy") = 1` browse for an archive in the default directory

The next two functions do not obey the normal Prode convention regarding the result returned if successful. A result of 0 means these were successful. They must be evaluated (followed by "=") in order for the operation to take place.

`mc_setErrMsg(0) = 0` set to 0 at start of calc's to clear flag

`wopt := 0`

`mc_defErrMsg(wopt) = 0` wopt = 0 turns off the Window Dialog messages  
wopt = 1 turns on the Window Dialog messages

## Open Properties window to view edit streams

`stream := 2`

`mc_edS(stream) = 1`

edit the given stream in the current active archive using the Prode window

`mc_edSS("dummy") = 1` ■

open to the first stream (disabled for the test to reduce popups)

## Chemical file operations

`mc_getFCNr("dummy") = 58`

number of components in data file, should be 1635 or greater. If less, then you are using the free version of Prode and not all of the routines below will work.

`id := 7732185`

CAS number of water (use internet search to find values for compounds)

compcode is an integer from 1 to number of components in the data file

compcode := mc\_CompCID(id) = 21

given id=CAS#, return compcode  
from database

Note: The above statement shows that the functions may be used to define a variable in addition to merely showing a result.

mc\_CompF(compcode) = "H2O"

given a component code, returns  
component formula string

mc\_CompN(compcode) = "WATER"

component name

mc\_CompID(compcode) = 7732185

CAS number of component, compare  
to id above

Note: The units are not returned by the Prode commands. Operations that show which units are being used are shown later.

mc\_CompMw(compcode) = 18.015

molecular weight

$T_c := mc\_CompTc(compcode) \cdot K = 647.096 K$

critical temperature

$$T_c = 1.165 \times 10^3 \cdot R$$

multiply the function by the current  
Prode units for the result to use the  
unit features of Mathcad

$mc\_CompPc(compcode) = 2.206 \times 10^7$

critical pressure

$mc\_CompVc(compcode) = 3.106 \times 10^{-3}$

critical volume

$mc\_CompAc(compcode) = 0.344$

acentric factor

$mc\_CompDm(compcode) = 0$

dipole moment

$mc\_CompRg(compcode) = 6.15 \times 10^{-11}$

radius of gyration

$mc\_CompSol(compcode) = 1.512 \times 10^3$

solubility parameter

$mc\_CompHf(compcode) = -1.342 \times 10^4$

heat of formation

$mc\_CompGf(compcode) = -1.27 \times 10^4$

Gibbs energy of formation

$mc\_CompSf(compcode) = 333.474$

enthalpy of fusion

$mc\_CompNb(compcode) = 373.15$

normal boiling point

$mc\_CompMp(compcode) = 273.15$

melting point temperature

The following provide non zero values only if the phase of interest is present at the temperature requested.

$tgl := 300$

temperature for gas/liquids (above  
freezing for water)

$ts := 260$	temperature for solids (below freezing)
$mc\_CompVP(compcode, tgl) = 3.548 \times 10^3$	saturation pressure at temp tgl
$mc\_CompHV(compcode, tgl) = 2.436 \times 10^3$	heat of vaporization at tgl
$mc\_CompLV(compcode, tgl) = 8.562 \times 10^{-4}$	liquid viscosity at tgl
$mc\_CompGV(compcode, tgl) = 9.925 \times 10^{-6}$	gas viscosity at tgl
$mc\_CompLD(compcode, tgl) = 995.476$	liquid density at tgl
$mc\_CompSD(compcode, ts) = 918.631$	solid density at ts
$mc\_CompLC(compcode, tgl) = 0.616$	liquid thermal conductivity at tgl
$mc\_CompGC(compcode, tgl) = 0.019$	gas thermal conductivity at tgl
$mc\_CompSC(compcode, ts) = 0$	solid thermal conductivity at ts (appears to be missing for water)
$mc\_CompST(compcode, tgl) = 0.072$	surface tension at tgl

integrated changes between two temperatures, t0 and t1 for pure components

$t0 := 280$	$t1 := 290$	
$mc\_CompHG(compcode, t0, t1) = 18.611$		ideal gas enthalpy change
$mc\_CompSG(compcode, t0, t1) = 0.065$		ideal gas entropy change
$mc\_CompHL(compcode, t0, t1) = 42.018$		ideal liquid enthalpy change
$mc\_CompSL(compcode, t0, t1) = 0.147$		ideal liquid entropy change
$ts0 := 260$	$ts1 := 270$	lower the temperature range < freezing pt
$mc\_CompHS(compcode, ts0, ts1) = 20.524$		ideal solid enthalpy change
$mc\_CompSS(compcode, ts0, ts1) = 0.077$		ideal solid entropy change

## Units commands

See "Units of Measurement" section in Prode manual for a list of the units and their numerical codes.

$UM := 15$	pressure is used for an example
$n\_press := mc\_getUMN(UM) = 20$	no. of units avail. for UM

mc\_getUMC(UM) = 1

present units code for UM

mc\_getSUMS(UM) = "Pa.a"

present units string for UM

sel := 5

select unit 5

mc\_getUMS(UM, sel) = "KPa.a"

units string for (UM, sel)

list all of the units for pressure

i := 1 .. n\_press

P\_units<sub>i</sub> := mc\_getUMS(UM, i)

	0
0	0
1	"Pa.a"
2	"Pa.g"
3	"mbar.a"
4	"mbar.g"
5	"KPa.a"
6	"KPa.g"
7	"bar.a"
8	"bar.g"
9	"kgf/cm <sup>2</sup> .a"
10	"kgf/cm <sup>2</sup> .g"
11	"psi.a"
12	"psi.g"
13	...

mc\_getP(stream)·Pa = 14.696·psi

multiply by current Prode pressure unit, then request any unit in the result

sel := 11

select a new pressure unit

mc\_setUMC(UM, sel) = 1

change to the 11th unit for pressure

mc\_getSUMS(UM) = "psi.a"

show current unit name for UM

mc\_getP(stream)·psi = 14.697·psi

now pressure results must be multiplied by psi

mc\_setUMC(UM, 1) = 1

reset to original unit for remainder of worksheet

Routines UMCR, UMCS, and UMAU are not fully documented in the Prode manual so they have been left out of the dll.

mc\_UMRAU(UM) = 1

removes all added units for (property no.)

## Error message flags

mc\_ErrMsg("") = "Error accessing component's data archive" last error message, maybe from a previous run

dum := "dummy"

errflag := mc\_getErrFlag(dum)

0 = no errors, 1 = errors found

errflag = 1

This flag only works if the Window Dialog messages are turned off. Otherwise, the Dialog messages are themselves the indication of errors. See mc\_defErrMsg. Errors that Mathcad detects (i.e. red indication) are not included for this flag.

At the time this test file was created, the thermal conductivity of solid water was not available in the database, causing an error and a value of 1 for errflag.

mc\_setErrFlag(0) = 0

set to 0 at start of next calc's to clear flag

mc\_defErrMsg(0) = ■ ■

0 = turns off the Window Dialog messages  
1 = turns on the Window Dialog messages

This function was demonstrated at start of worksheet. Turning off the Window Dialog messages allows the computations to continue without pausing to close the Dialog window when an error occurs. The error may still be visible if a 0 value is returned where a real number is expected.

## Atmospheric pressure

patm := mc\_getPatm("mc") =  $1.013 \times 10^5$

the pressure should be  $1.013 \cdot 10^5$

## Base values for enthalpy and entropy

This section is new for this version

The default values for the base temperature, enthalpy, and entropy may be found in the config>settings Prode window. The functions below may be used to change the settings. The settings apply only to the current archive. The settings for the archive are saved when the archive is saved.

### Code Procedure

1 = initial values specified by user (values of tref and val)

2 = initial values are enthalpy of formation (or entropy of formation) and temperature 25 C

If code = 2, the tref and val inputs are ignored.

code := 1                      tref := 298                      val := 0

mc\_setHB(code, tref, val) = 1                      enthalpy references

mc\_setSB(code, tref, val) = 1                      entropy references

## Read/write stream properties

If a write operation exists, it will appear under the read operation, using the value from the read operation. This simplifies the testing process.

The write operations in this section are in blue highlight.

stream := 1

phase := 2                      the phase position (not the phase type)

cpos := 2                      cpos is the component's numerical position in the composition vector for the stream, starting with 1

mc\_isSDef (stream) = 1                      given a stream returns TRUE (integer = 1) if stream has been defined, otherwise returns FALSE (0)

name := mc\_StrN(stream) = "Test Case 1"                      stream name

mc\_putN(stream, name) = 1

mc\_setOp(stream, 150, patm) = 1

This is an edit operation to lower the temperature so liquid will be present for the functions below.

t := mc\_getT(stream) = 150                      temperature

mc\_putT(stream, t) = 1

$p := mc\_getP(stream) = 1.013 \times 10^5$

pressure

$mc\_putP(stream, p) = 1$

$pnr := mc\_getPNr("mc") = 5$

returns the maximum number of phases that procedure can detect in the archive for all streams (may include phases at other temperatures)

$mc\_StrPt(stream, phase) = 1$

given a stream and phase # in range 1-  
getPNr() returns the phase type  
(0=vapor, 1=liquid, 2=solid)

$i := 0..pnr - 1$

$phases_i := mc\_StrPts(stream, i + 1)$

given a stream and phase # in range 1-  
getPNr() returns a ANSI C string with  
the description of type for detected  
phase

$$phases = \begin{pmatrix} \text{"Vapor"} \\ \text{"Liquid"} \\ \text{"Liquid"} \\ \text{"Not present"} \\ \text{"Not present"} \end{pmatrix}$$

only one liquid phase is present  
later, the flash routine code will be  
reset to obtain two liquid phases

$mc\_StrLf(stream) = 0.26$

given a stream returns the total liquid  
fraction (molar basis) in stream

$mc\_StrPf(stream, phase) = 0.161$

given a stream and phase # in  
range 1- getPNr() returns the phase  
fraction

$w := mc\_getW(stream, phase, cpos) = 0.272$

mole fraction of component (cpo #)  
in a phase

$mc\_putW(stream, phase, cpos, w) = 1$

mole fraction w of component cpos in  
a phase

$rate := mc\_getWm(stream) = 1$

stream flow rate, mass/time

$mc\_setWm(stream, rate) = 1$

$zi := mc\_getZ(stream, cpos) = 0.15$

mole fraction of component cpos in  
total stream

$mc\_putZ(stream, cpos, zi) = 1$



<code>mc_getCNR(stream) = 3</code>	number of components in stream
<code>mc_StrZv(stream) = 0.984</code>	returns the relevant compressibility factor (gas phase)
<code>mc_StrMw(stream) = 22.944</code>	molecular weight of total stream
<code>mc_StrGMw(stream) = 17.563</code>	molecular weight of gas phase
<code>mc_StrLMw(stream) = 38.272</code>	molecular weight of liquid phase
<code>mc_StrV(stream) = 0.391</code>	specific volume as sum of specific volumes of all phases

### ***enthalpy***

<code>h := mc_StrH(stream) = -434.413</code>	total stream enthalpy
<code>mc_StrGH(stream) = -163.905</code>	gas phase enthalpy
<code>mc_StrSGH(stream) = -289.294</code>	gas specific enthalpy
<code>mc_StrLH(stream) = -270.508</code>	liquid enthalpy
<code>mc_StrSLH(stream) = -624.11</code>	liquid specific enthalpy
<code>mc_StrSH(stream) = 0</code>	solid enthalpy
<code>mc_StrSSH(stream) = 0</code>	solid specific enthalpy

### ***entropy***

<code>entropy := mc_StrS(stream) = -2.137</code>	total stream entropy
<code>mc_StrGS(stream) = -0.756</code>	gas phase entropy
<code>mc_StrSGS(stream) = -1.335</code>	gas specific entropy
<code>mc_StrLS(stream) = -1.38</code>	liquid entropy
<code>mc_StrSLS(stream) = -3.185</code>	liquid specific entropy
<code>mc_StrSS(stream) = 0</code>	solid entropy
<code>mc_StrSSS(stream) = 0</code>	solid specific entropy

### ***heat capacity, mass basis***

<code>mc_StrGICp(stream) = 1.887</code>	ideal gas heat capacity
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$$\text{mc\_StrGCp}(\text{stream}) = 1.915$$

gas constant pressure heat capacity

$$\text{mc\_StrGCv}(\text{stream}) = 1.416$$

gas constant volume heat capacity

$$\text{mc\_StrLCp}(\text{stream}) = 1.737$$

liquid constant pressure heat capacity

$$\text{mc\_StrLCv}(\text{stream}) = 1.215$$

liquid constant volume heat capacity

$$\text{mc\_StrSCp}(\text{stream}) = 0$$

solid constant pressure heat capacity

### ***speed of sound***

$$\text{mc\_StrMSS}(\text{stream}) = 0$$

mixed phase speed of sound HEM model

$$\text{mc\_StrGSS}(\text{stream}) = 266.864$$

gas phase

$$\text{mc\_StrLSS}(\text{stream}) = 1.894 \times 10^3$$

liquid phase

### ***Joule Thomson coefficient***

$$\text{mc\_StrGJT}(\text{stream}) = 1.528 \times 10^{-5}$$

gas phase

$$\text{mc\_StrLJT}(\text{stream}) = -3.992 \times 10^{-7}$$

liquid phase

### ***compressibility, expansivity***

$$\text{mc\_StrGIC}(\text{stream}) = 1.003 \times 10^{-5}$$

gas isothermal compressibility,  
 $\frac{1}{V} \cdot \left( \frac{d}{dP} V \right)$

$$\text{mc\_StrLIC}(\text{stream}) = 5.977 \times 10^{-10}$$

liquid isothermal compressibility

$$\text{mc\_StrGVE}(\text{stream}) = -6.949 \times 10^{-3}$$

gas volumetric expansivity  $\frac{1}{V} \cdot \left( \frac{d}{dT} V \right)$

$$\text{mc\_StrLVE}(\text{stream}) = -1.522 \times 10^{-3}$$

gas volumetric expansivity

### ***density***

$$\text{mc\_StrGD}(\text{stream}) = 0 \times 10^0$$

gas density

$$\text{mc\_StrLD}(\text{stream}) = 1.113 \times 10^3$$

liquid density

### ***thermal conductivity***

$$\text{mc\_StrGC}(\text{stream}) = 0.015$$

gas conductivity

$$\text{mc\_StrLC}(\text{stream}) = 0.285$$

liquid conductivity

### ***viscosity***

$$\text{mc\_StrGV}(\text{stream}) = 6.261 \times 10^{-6}$$

gas viscosity

$$\text{mc\_StrLV}(\text{stream}) = 1.649 \times 10^{-4}$$

liquid viscosity

### ***surface tension***

$$\text{mc\_StrST}(\text{stream}) = 0.034$$

liquid/gas

### ***flammability***

$$\text{mc\_StrFML}(\text{stream}) = 4.993$$

gas phase lean limit

$$\text{mc\_StrFMH}(\text{stream}) = 15.082$$

gas phase rich limit

### ***other stream properties***

$$\text{mc\_StrHC}(\text{stream}) = 4.332 \times 10^4$$

gas phase heat of combustion

$$\text{compcode} := \text{mc\_getCC}(\text{stream}, \text{cpos}) = 2$$

component number for component =  
cpos

$$\text{mc\_putCC}(\text{stream}, \text{cpos}, \text{compcode}) = 1$$

$$\text{mc\_getMCNr}(\text{"dummy"}) = 50$$

maximum number of components in a  
stream

### ***interactions***    This section was changed for this version.

$$\text{int\_pos} := 1$$

the interaction number (i.e. the row in  
the BIP window for the stream)

$$\text{mc\_getMBPNr}(\text{"dummy"}) = 250$$

maximum number of binary pairs in a  
stream (for all streams)

$$\text{ci} := \text{mc\_getCi}(\text{stream}, \text{int\_pos}) = 1$$

component index i in interaction list

$$\text{mc\_putCi}(\text{stream}, \text{int\_pos}, \text{ci}) = 1$$

$$\text{cj} := \text{mc\_getCj}(\text{stream}, \text{int\_pos}) = 2$$

component index j in interaction list

$$\text{mc\_putCj}(\text{stream}, \text{int\_pos}, \text{cj}) = 1$$

$$\text{model} := \text{mc\_getMB}(\text{stream}, \text{int\_pos}) = 50$$

returns model number for interaction  
# int\_pos in given stream

$$\text{mc\_putMB}(\text{stream}, \text{int\_pos}, \text{model}) = 1$$

sets the model number for a given  
stream and interaction number,  
int\_pos

id := 0

This id is the BIP column (starting with 0 for BIP-1) shown in the PPP window under BIPs for the stream.

Kji := mc\_getBIP(stream,int\_pos,id) = 0.08

value of the interaction coefficient

mc\_putBIP(stream,int\_pos,id,Kji) = 1

specifies a value for an interaction coefficient

### ***thermodynamic models for streams***

This section was changed for this version.

stream = 1

Kcode := 50

code for SRK standard model package, see manual for other package numbers

mc\_setKM(stream,Kcode) = 1

mp codes for functions below

- 1 fugacity
- 2 enthalpy
- 3 entropy
- 4 molar volume
- 5 viscosity

Examples for fugacity and enthalpy models below:

state := 0

vapor state

mp := 1

fugacity model for stream, state

fmodel := mc\_getMP(stream,mp,state) = 50

retrieve model number

mc\_setMP(stream,mp,fmodel,state) = 1

set model number

mp := 2

enthalpy model for stream, state

hmodel := mc\_getMP(stream,mp,state) = 50

retrieve model number

mc\_setMP(stream,mp,hmodel,state) = 1

set model number

## Thermodynamic calculations

stream := 5

t = 150

p =  $1.013 \times 10^5$

state := 1

use stream 5 for examples below

state (0=vapor, 1=liquid, 2=solid)

### ***phase equilibria***

n := 1

pf := .3

see below

mc\_PfPF(stream, p, pf, state, n) = 0

n th equilibrium temp at p, pf (phase fraction), state (0=vapor, 1=liquid, 2=solid)

mc\_PfTF(stream, t, pf, state, n) = 0

n th equil. press at t, pf, state

lf := .2

set liquid fraction

mc\_LfPF(stream, p, lf) = 300.674

first equil. temp at liquid fraction, lf

mc\_LfTF(stream, t, lf) = 0

first equil. pressure at liquid fraction, lf

mc\_StrCPnr(stream) = 1

number of critical points found

cpn := 1

selected critical point

mc\_StrPc(stream, cpn) =  $5.242 \times 10^6$

critical pressure for critical point #, cpn

mc\_StrCBp(stream) = 0

cricondenBar pressure

mc\_StrCBt(stream) = 0

cricondenBar temperature

mc\_StrCTp(stream) =  $4.982 \times 10^6$

cricondenTherm pressure

mc\_StrCTt(stream) = 457.544

cricondenTherm temperature

mc\_StrAc(stream) = 0.208

acentric factor (mole fraction average)

### phase diagrams

stream := 5

lnr := mc\_PELNr(stream)

Given a stream calculates the phase diagram and returns the number of equilibrium lines available

lnr = 2

### line types

line := 2

ltype := mc\_PELT(stream, line)

ltype = 2

Given a stream and line number,  
returns the line type:

1. bubble line
2. dew line
3. three phase line
4. fractional phase

### line properties

lprop := mc\_PELP(stream, line)

lprop = 1

Given a stream and line, returns the  
line properties:

1. vapor-liquid
2. vapor-liquid-liquid
3. vapor-solid
4. liquid-solid
5. fractional phase

### equilibrium lines

The prode.dll has assumed a maximum number of points of 50 for the equilibrium lines. This dimension cannot be changed dynamically for the variables passed to and from Mathcad. Therefore, the mc\_PELine routine leaves out the maxpt variable that is shown in the Prode corresponding routine.

The mc\_PELine function (see the first line in the program below) produces a matrix result. Although this matrix can be used "as is" the Mathcad program, "PELine" below calls mc\_PELine and splits the matrix into the separate variables.

```
PELine(stream, line) := 
$$\left| \begin{array}{l} M \leftarrow mc\_PELine(stream, line) \\ npts \leftarrow M_{0,2} \\ T \leftarrow submatrix(M, 0, npts - 1, 0, 0) \\ P \leftarrow submatrix(M, 0, npts - 1, 1, 1) \\ (T \quad P \quad npts) \end{array} \right|$$

```

(T1 P1 npts1) := PELine(stream, line)

The output is shown below.

npts1 = 33

Given stream and equilibrium line  
number, the temperature and  
pressure vectors and the total  
number of points are computed and  
returned.

	0
0	311.885
1	320.94
2	325.94
3	330.94
4	335.94
5	340.94
6	345.94
7	350.94
8	355.94
9	360.94
10	365.94
11	370.94
12	375.94
13	...

T1 =

	0
0	$1.013 \cdot 10^5$
1	$1.415 \cdot 10^5$
2	$1.686 \cdot 10^5$
3	$1.997 \cdot 10^5$
4	$2.351 \cdot 10^5$
5	$2.754 \cdot 10^5$
6	$3.21 \cdot 10^5$
7	$3.723 \cdot 10^5$
8	$4.299 \cdot 10^5$
9	$4.944 \cdot 10^5$
10	$5.663 \cdot 10^5$
11	$6.462 \cdot 10^5$
12	$7.348 \cdot 10^5$
13	...

P1 =

### phase fraction lines

stream := 5

state := 0

fraction := .5

PFLine(stream, state, fraction) :=

M ← mc_PFLine(stream, state, fraction)	
npts ← M <sub>0,2</sub>	
T ← submatrix(M, 0, npts - 1, 0, 0)	
P ← submatrix(M, 0, npts - 1, 1, 1)	
(T P npts)	

(Tf Pf nf) := PFLine(stream, state, fraction)

nf = 37

Given stream, state, and fraction of that state, computes the temperature and pressure vectors along that phase fraction, plus the number of points on the curve.

	0
0	272.895
1	277.895
2	282.895
3	287.895
4	292.895
5	297.895
Tf = 6	302.895
7	307.895
8	312.895
9	317.895
10	322.895
11	327.895
12	332.895
13	...

	0
0	$1.013 \cdot 10^5$
1	$1.21 \cdot 10^5$
2	$1.436 \cdot 10^5$
3	$1.693 \cdot 10^5$
4	$1.985 \cdot 10^5$
5	$2.313 \cdot 10^5$
Pf = 6	$2.683 \cdot 10^5$
7	$3.096 \cdot 10^5$
8	$3.556 \cdot 10^5$
9	$4.066 \cdot 10^5$
10	$4.63 \cdot 10^5$
11	$5.251 \cdot 10^5$
12	$5.933 \cdot 10^5$
13	...

The Mathcad program, "PhaseEnv" below obtains all of the equilibrium curves which can then be plotted.

```

PhaseEnv(stream) :=
    lnr ← mc_PELNr(stream)
    for line ∈ 1..lnr
         $\left( \begin{matrix} \langle t \rangle_{line-1} & \langle p \rangle_{line-1} & n_{line-1} \end{matrix} \right) \leftarrow PELine(stream, line)$ 
        ltypeline-1 ← mc_PELT(stream, line)
        lpropline-1 ← mc_PELP(stream, line)
    (t p lnr n ltype lprop)

```

stream := 5

(Tj Pj lnr nc type prop) := PhaseEnv(stream)

lnr = 2

$type = \begin{pmatrix} 1 \\ 2 \end{pmatrix}$ 
 $nc = \begin{pmatrix} 43 \\ 33 \end{pmatrix}$

As shown in the nc vector, the lines may have different number of points. In order to prevent curves returning to the origin, extract the data from Tj and Pj.

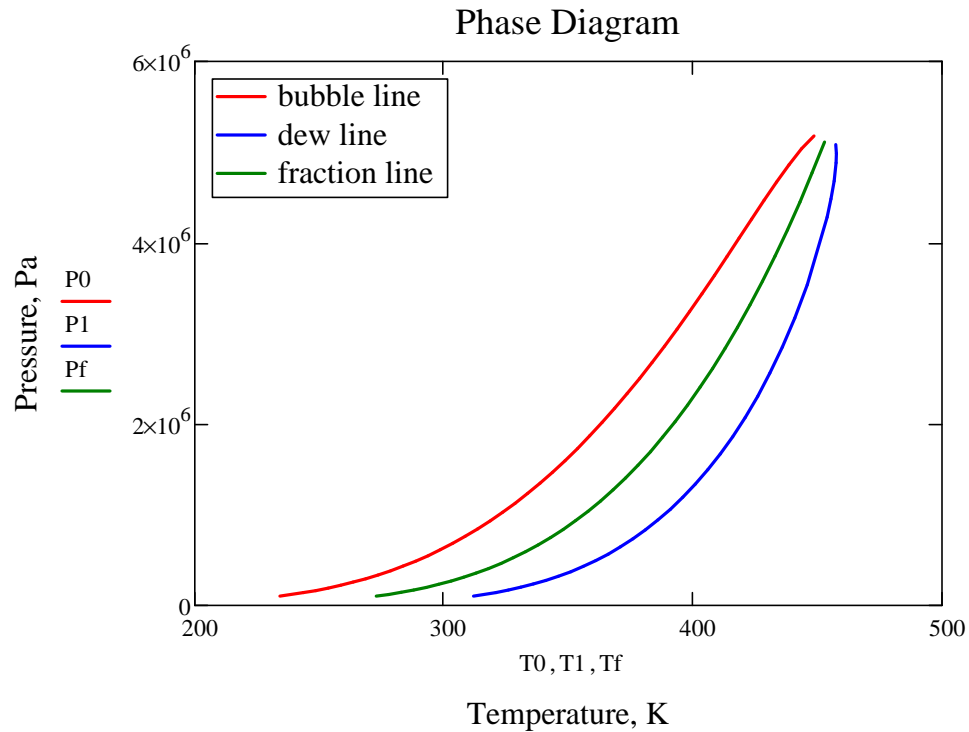


$T0 := \text{submatrix}(Tj, 0, nc0 - 1, 0, 0)$

$P0 := \text{submatrix}(Pj, 0, nc0 - 1, 0, 0)$

$T1 := \text{submatrix}(Tj, 0, nc1 - 1, 1, 1)$

$P1 := \text{submatrix}(Pj, 0, nc1 - 1, 1, 1)$



The legend labels in the above plot were input manually supplied based on the values of the "type" results. Also, the number of curves is determined manually using `Inr` as guidance. Some streams have more equilibrium curves due to multiple liquid and/or solids.

The `PELine`, `PFLine`, and `PhaseEnv` programs may be copied into or referenced by other programs.

## hydrates

`hydmodel := 1`      `thyd := 260`

`str_hyd := 2`      stream for hydrate function below

`hydmodel =`

1 = assume free water present, this option produces conservative but safe values

2 = calculate amount of water in liquid phase

3 = solve as multiphase equilibria, solve phase equilibria including solids as ice

Since water is not present in the stream chosen for testing, the 2 and 3 hydmodels will return 0.

$\text{phyd} := \text{mc\_HPFORM}(\text{str\_hyd}, \text{thyd}, \text{hydmodel}) = 1.821 \times 10^6$  returns the pressure that hydrates form at temperature = thyd

The HTFORM Prode function is not available in the Basic version, but the HPFORM function should suffice.

## ***flashes***

$\text{stream} = 5$

$\text{mc\_setSOp}(\text{stream}) = 1$

flash at standard conditions

$\text{et} := 0$

estimated temperature set to 0 for automatic

$\text{ep} := 0$

estimated pressure set to 0 for automatic

$\text{mc\_setOp}(\text{stream}, 150, \text{patm}) = 1$

set new operating conditions and flash

$\text{t} := \text{mc\_getT}(\text{stream}) = 150$

temperature

$\text{p} := \text{mc\_getP}(\text{stream}) = 1.013 \times 10^5$

pressure

$\text{h} := \text{mc\_StrH}(\text{stream}) = -697.466$

enthalpy obtained above

$\text{entropy} := \text{mc\_StrS}(\text{stream}) = -2.866$

entropy obtained above

$\text{sv} := \text{mc\_StrV}(\text{stream}) = 1.405 \times 10^{-3}$

volume obtained above

### **find temperature**

$\text{mc\_VPF}(\text{stream}, \text{p}, \text{sv}, \text{et}) = 150$

volume-pressure flash, et=temp guess

$\text{mc\_HPF}(\text{stream}, \text{p}, \text{h}, \text{et}) = 0$

enthalpy-pressure flash, et=temp guess

$\text{mc\_SPF}(\text{stream}, \text{p}, \text{entropy}, \text{et}) = 0$

entropy-pressure flash, et=temp guess

### **find pressure**

$\text{mc\_VTF}(\text{stream}, \text{t}, \text{sv}, \text{ep}) = 0$

volume-temp flash, ep=press guess

$\text{mc\_HTF}(\text{stream}, \text{t}, \text{h}, \text{ep}) = 0$

enthalpy-temp flash, ep=press guess

$\text{mc\_STF}(\text{stream}, \text{t}, \text{entropy}, \text{ep}) = 0$

entropy-temp flash, ep=press guess

The flashes that determine pressure have some difficulty converging for multiphase (liquids and solids) problems. Select another flash routine and iterate if needed.

Additional flashes for mixing and dividing streams are found at this [section](#)

## Extended methods for accessing stream properties

These functions allow simultaneous setting of temperature and pressure followed by an isothermal flash before the desired property is returned. These methods should be used with care because of the change in the stream conditions.

stream := 2

mc\_EStrGMw(stream,t,p) = 16.047

gas molecular weight

mc\_EStrLMw(stream,t,p) = 55.298

liquid molecular weight

mc\_EStrLf(stream,t,p) =  $9.572 \times 10^{-4}$

mole fraction of liquid

phase := 1

position of phase, not the state code.  
positions are usually vapor=1,  
liquid=2, solid=3 but extra liquid and  
solid phases may be present

mc\_EStrPf(stream,phase,t,p) = 0.999

molar phase fraction of phase

mc\_EStrZv(stream,t,p) = 0.984

gas (vapor) compressibility factor

mc\_EStrH(stream,t,p) = -317.84

total enthalpy

mc\_EStrV(stream,t,p) = 0.752

total specific volume

mc\_EStrGCp(stream,t,p) = 2.103

gas constant pressure heat capacity

mc\_EStrGCv(stream,t,p) = 1.559

gas constant volume heat capacity

mc\_EStrLCp(stream,t,p) = 1.709

liquid constant pressure heat capacity

mc\_EStrLCv(stream,t,p) = 1.399

liquid constant volume heat capacity

mc\_EStrGIC(stream,t,p) =  $1.003 \times 10^{-5}$

gas isothermal compressibility

mc\_EStrLIC(stream,t,p) =  $5.211 \times 10^{-10}$

liquid isothermal compressibility

mc\_EStrMSS(stream,t,p) = 0

mixture speed of sound

mc\_EStrGSS(stream,t,p) = 0

gas speed of sound

mc\_EStrLSS(stream,t,p) =  $3.325 \times 10^3$

liquid speed of sound

$mc\_EStrGJT(stream, t, p) = 1.466 \times 10^{-5}$	gas Joule Thomson coefficient
$mc\_EStrLJT(stream, t, p) = -6.958 \times 10^{-7}$	liquid Joule Thomson coefficient
$mc\_EStrGVE(stream, t, p) = -6.939 \times 10^{-3}$	gas volumetric expansivity coefficient
$mc\_EStrLVE(stream, t, p) = -8.86 \times 10^{-4}$	liquid volumetric expansivity coefficient
$mc\_EStrHC(stream, t, p) = 5.001 \times 10^4$	heat of combustion
$mc\_EStrFML(stream, t, p) = 4.999$	lean flammability limit of gas
$mc\_EStrFMH(stream, t, p) = 14.999$	rich flammability limit of gas
$mc\_EStrS(stream, t, p) = -1.466$	total entropy
$mc\_EStrGD(stream, t, p) = 1.325$	gas density
$mc\_EStrLD(stream, t, p) = 729.12$	liquid density
$mc\_EStrGC(stream, t, p) = 0.016$	gas thermal conductivity
$mc\_EStrLC(stream, t, p) = 0.175$	liquid thermal conductivity
$mc\_EStrGV(stream, t, p) = 6.028 \times 10^{-6}$	gas viscosity
$mc\_EStrLV(stream, t, p) = 1.245 \times 10^{-3}$	liquid viscosity
$mc\_EStrST(stream, t, p) = 0.029$	surface tension

## Fugacity and derivatives

This section was changed for this version.

The operations below behave like subroutines rather than functions because they return more than one result. The Mathcad system imposes some restrictions on function input and output so the normal C++ methods of passing variables is not possible. These restrictions are needed to enforce the "non code" look of the Mathcad interface. As a result of these restrictions, the functions below have slightly different argument lists than found in Prode and all of the results are returned in a single matrix. Mathcad routines are then provided to split these results into the appropriate variables.

The prode.dll has assumed a maximum number of components of 50 for all vector and matrix routines. This dimension cannot be changed dynamically for the variables passed to and from Mathcad. For greater number of components, prode.dll must be rebuilt. The constant "maxnc" in the source code for the routines in this section must be changed to the higher number.

```
stream := 1
```

```
NC := mc_getCNR(stream)
```

t = 150

These variables were defined above.

p =  $1.013 \times 10^5$

mc\_setOp(stream,t,p) = 1

i := 0..NC - 1

phase := 2

w<sub>i</sub> := mc\_getW(stream,phase,i + 1)

$$w = \begin{pmatrix} 6.329 \times 10^{-3} \\ 0.272 \\ 0.722 \end{pmatrix}$$

state := 1

The liquid state is being used.

New in version 1.2b: The fugacity, H, S, and V functions and their derivatives previously included the stream number as the first argument. Now the first argument is a "process code". The code is set using the mc\_DPinit function which also loads the stream information into active memory. The stream (process) stays loaded, making execution of StrFv and similar routines faster. This is especially useful if the routines are repeatedly called by a program loop.

process := 1

Up to 5 processes may be defined in the base Prode version. Processes may be redefined with new streams.

mc\_DPinit(process,stream) = 1

### fugacity vector

fg := mc\_StrFv(process,state,t,p,w,NC)·Pa

This routine returns the fugacity vector alone.

$$fg = \begin{pmatrix} 1.488 \times 10^7 \\ 1.788 \times 10^4 \\ 1.056 \times 10^3 \end{pmatrix} \text{ Pa}$$

fugacity<sub>i</sub> := fg<sub>i</sub>·w<sub>i</sub>

The Prode routines define the "fugacity" variable as the fugacity coefficient times the

$$\text{fugacity} = \begin{pmatrix} 9.416 \times 10^4 \\ 4.86 \times 10^3 \\ 762.147 \end{pmatrix} \text{ Pa}$$

total pressure. Thus, fugacity is obtained by the equation on the left.

### fugacity vector plus derivatives wrt T, P, w

```
StrFvd(process, state, t, p, w, NC) :=
  M ← mc_StrFvd(process, state, t, p, w, NC)
  "separate the results into vectors and a matrix"
  fg ← submatrix(M, 0, NC - 1, 0, 0)
  dfgt ← submatrix(M, 0, NC - 1, 1, 1)
  dfgp ← submatrix(M, 0, NC - 1, 2, 2)
  dfgw ← submatrix(M, 0, NC - 1, 3, 3 + NC - 1)
  (fg dfgt dfgp dfgw)
```

(fg dfgt dfgp dfgw) := StrFvd(process, state, t, p, w, NC) given the stream, state, temp, press, composition vector, w, and the number of components, NC, return fugacity vector, fg, and derivatives of fg wrt t, p, and w.

Add the default Prode units as needed.

$$\text{fg} \cdot \text{Pa} = \begin{pmatrix} 1.488 \times 10^7 \\ 1.788 \times 10^4 \\ 1.056 \times 10^3 \end{pmatrix} \text{ Pa} \quad \text{dfgt} \cdot \frac{\text{Pa}}{\text{K}} = \begin{pmatrix} 2.686 \times 10^5 \\ 1.627 \times 10^3 \\ 115.318 \end{pmatrix} \cdot \frac{\text{Pa}}{\text{K}} \quad \text{dfgp} = \begin{pmatrix} 2.686 \times 10^5 \\ 1.627 \times 10^3 \\ 115.318 \end{pmatrix}$$

$$\text{dfgw} \cdot \text{Pa} = \begin{pmatrix} -2.139 \times 10^8 & -1.533 \times 10^8 & -1.202 \times 10^8 \\ -6.41 \times 10^4 & -7.652 \times 10^4 & -2.018 \times 10^4 \\ 1.557 \times 10^3 & 1.796 \times 10^3 & 523.217 \end{pmatrix} \text{ Pa}$$

## Other stream state variables and their derivatives

This section was changed for this version.

Functions were provided above (eg. mc\_StrH) to obtain the enthalpy (H), entropy (S), and molar volume (V) of a stream. The next routine allows the operating conditions (t, p, w) to be specified to values other than those in the stream data file. The user selects which variable, H, S, or V, is desired, using a string variable with the corresponding variable initial. The program calls the appropriate mc\_xxx function and then separates the variables from the output matrix.

```

StrXvd(X,process,state,t,p,w,NC) := 
    M ← mc_StrHvd(process,state,t,p,w,NC) if X = "H"
    M ← mc_StrSvd(process,state,t,p,w,NC) if X = "S"
    M ← mc_StrVvd(process,state,t,p,w,NC) if X = "V"
    x ← M(0)
    dxt ← M(1)
    dxp ← M(3)
    dxw ← submatrix(M,0,0,3,NC+2)
    (x dxt dxp dxw)

```

KJ := 1000·J

Kmol := 1000·mol

define new units for Mathcad

(H dHt dHp dHw) := StrXvd("H",process,state,t,p,w,NC)

$$H \cdot \frac{\text{KJ}}{\text{Kmol}} = (-2.419 \times 10^4) \cdot \frac{\text{KJ}}{\text{Kmol}}$$

the default Prode units have been applied to the results

$$dHt \cdot \frac{\frac{\text{KJ}}{\text{Kmol}}}{\text{K}} = (65.204) \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{K}}$$

$$dHp \cdot \frac{\frac{\text{KJ}}{\text{Kmol}}}{\text{Pa}} = (-8.027 \times 10^3) \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{Pa}}$$

$$dHw \cdot \frac{\text{KJ}}{\text{Kmol}} = (-8.027 \times 10^3 \quad -2.153 \times 10^4 \quad -2.533 \times 10^4) \cdot \frac{\text{KJ}}{\text{Kmol}}$$

(S dSt dSp dSw) := StrXvd("S",process,state,t,p,w,NC)

$$S \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{K}} = (-121.246) \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{K}}$$

$$dSt \cdot \frac{\frac{\text{KJ}}{\text{Kmol}}}{\text{K}^2} = (0.433) \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{K}^2}$$

$$dSp \cdot \frac{\frac{\text{KJ}}{\text{Kmol}}}{\text{K} \cdot \text{Pa}} = (-70.635) \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{K} \cdot \text{Pa}}$$

$$dSw \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{K}} = (-70.635 \quad -105.28 \quad -104.679) \cdot \frac{\text{KJ}}{\text{Kmol} \cdot \text{K}}$$

$$(V \quad dVt \quad dVp \quad dVw) := \text{StrXvd}("V", \text{process}, \text{state}, t, p, w, \text{NC})$$

$$V \cdot \frac{\text{m}^3}{\text{mol}} = (0.034) \frac{\text{m}^3}{\text{mol}}$$

$$dVt \cdot \frac{\text{m}^3}{\text{mol} \cdot \text{K}} = (4.845 \times 10^{-5}) \frac{\text{m}^3}{\text{mol} \cdot \text{K}}$$

$$dVp \cdot \frac{\text{m}^3}{\text{mol} \cdot \text{Pa}} = (0.073) \cdot \frac{\text{m}^3}{\text{mol} \cdot \text{Pa}}$$

$$dVw \cdot \frac{\text{m}^3}{\text{mol}} = (0.073 \quad 0.069 \quad 0.068) \frac{\text{m}^3}{\text{mol}}$$

## Operations to set/retrieve the options needed for equation of state models and flash routine phases

See the Prode manual (see paragraph "Codes used in Prode library") and also open the Prode drop menus for the model to view the description of the options set by the OM code variable. The user will probably find it easier to set the options using the Prode window.

All even code values mean that only single liquid phases are allowed in the flash routines. For multiple liquids, the code value must be odd.

stream := 1

option := mc\_getOM(stream) = 545

current option set

This should = 552 for stream 1 in the def.ppp archive. With this option value, only L-V flashes will result.

mc\_setOM(stream, option + 1) = 1

This changes the flashes of "stream" to multiple liquids.

mc\_setOp(stream, t, p) = 1

redo the flash

i := 0..pnr - 1

given a stream and phase # in range  
1- getPNr() returns the phase type  
(0=vapor, 1=liquid, 2=solid)

phases<sub>i</sub> := mc\_StrPts(stream, i + 1)

given a stream and phase # in range  
1- getPNr() returns a ANSI C string  
with the description of type for  
detected phase



$$\text{phases} = \begin{pmatrix} \text{"Vapor"} \\ \text{"Solid"} \\ \text{"Solid"} \\ \text{"Not present"} \\ \text{"Not present"} \end{pmatrix}$$

Previously, only one liquid was present

`mc_setOM(stream,option) = 1`

reset the code to single liquids

## Initializing a stream

This section was changed for this version.

The example will create a stream with water and methanol. The component numbers in the Prode databank can change with updates, so always use CAS numbers when initializing by program instead of manually using the Prode window.

`methanol_id := 67561`

CAS number of methanol

`methanol_code := mc_CompCID(methanol_id) = 24`

`water_id := 7732185`

CAS number of water

`water_code := mc_CompCID(water_id) = 21`

`stream := 11`

`mc_initS(stream) = 1`

initialize a new stream

`model := 50`      SRK standard

see Prode manual for model codes

`mc_setKM(stream,model) = 1`

set property model package

`mc_putZ(stream,1,.5) = 1`

set total stream mole fractions

`mc_putZ(stream,2,.5) = 1`

`mc_putCC(stream,1,methanol_code) = 1`

define components

`mc_putCC(stream,2,water_code) = 1`

`mc_setS(stream) = 1`

validate the stream

`mc_loadSB(stream) = 1`

load BIP coefficients

`mc_setWm(stream,1.3) = 1`

set mass flow rate

`temp := 300`      `pres := patm`

`mc_setOp(stream,temp,pres) = 1`

set temp and pres and flash

`mc_edS(stream) = 0`

view the stream then press OK

## Other stream operations

stream2 := 1

stream1 := 10

mc\_StrCopy(stream1, stream2) = 1

copy stream2 to stream1 (note the order!)

et := mc\_getT(stream2) = 150

mc\_getT(stream1) = 150

mc\_MixF(stream1, stream2, et) = 1

flash at lower stream press, et=temp guess for mixed stream. **The sum of the streams replaces stream1. A new stream is NOT created.**

mc\_getT(stream1) = 329.55

mixed stream temperature

stream2 := 12

the new stream to be created by Divi

wdiv := .7

mc\_Divi(stream1, stream2, wdiv) = 1

Given one stream (stream1) and a flowrate fraction (0-1) performs a divider operation so that stream 1 is shifted into two streams (stream1, stream2) of the same composition, temperature and pressure, flowrate fractions are subdivided as specified by wdiv (stream2 = wdiv, stream1 = 1- wdiv)

**Only one new stream is created, NOT two. The starting stream gets overwritten.**

## phase separation

stream1 := 5

stream2 := 13

the new stream to be created by PSep

phase := 1

phase number to separate, NOT the phase type

mc\_PSep(stream1 , stream2 , phase) = 1

Given a stream (stream1) performs an isothermal flash to simulate a phase separator and returns the specified phase number (not phase type) to stream2.

gasstream := 14

mc\_GSep(stream1 , gasstream) = 1

Given a stream (stream1) performs an isothermal flash to simulate a phase separator and returns the gas phase to gasstream

liqstream := 15

mc\_LSep(stream1 , liqstream) = 1

Given a stream (stream1) performs an isothermal flash to simulate a phase separator and returns the liq phase(s) to liqstream

## Polytropic compressor/expander

rate compressor efficiency

pin :=  $10^6$       pressure in Pa

pout :=  $2 \cdot 10^6$

tin := 300      temperature in K

tout := 370

model := 2

for a rating, model may be the following:

2 = Huntington method

4 = Paron method

stream := 2

mc\_setOp(stream , tin , pin) = 1

set the inlet stream conditions

mc\_PSPF(stream , pout , model , tout) = 0.743

efficiency rating and "stream" in archive now contains the outlet conditions

design model

eff := .75

polytropic efficiency given

model := 1

for a design, model may be the following:

1 = Huntington

3 = Paron

mc\_setOp(stream , tin , pin) = 1

reset the inlet stream conditions

mc\_PSPF(stream , pout , model , eff) = 369.276

outlet temperature and "stream" now contains the outlet conditions

## Isentropic expansion, nozzles

stream := 5

tin := 340              pin :=  $2 \cdot 10^6$

mc\_setWm(stream, 1.23) = 1

mc\_setOp(stream, tin, pin) = 1

set stream conditions

model := 2

model options::

1 = homogeneous, equilibrium

2 = homogeneous, non equilibrium

3 = homogeneous, non equilibrium ?

4 = nonhomogeneous, non  
equilibrium

pout := patm

parameter := .75

Prode manual does not explain

mc\_ISPF(stream, pout, model, parameter) = 0

calculated orifice area, m<sup>2</sup>

mc\_getErrFlag(" ") = 1

## Pipe flow

The PIPE function is only available for users with an extended Prode license.

model := 1

stream := 1

diam :=  $\frac{1 \cdot \text{in}}{\text{m}} = 0.025$

rough := .00045

length :=  $\frac{100 \cdot \text{km}}{\text{m}} = 1 \times 10^5$

dHeight := 0

dHeat := 0

mc\_PIPE(stream, model, diam, rough, length, dHeight, dHeat) = 0

The result above will be 0 if the user has a Basic Prode license or 1 for an Extended license. The pressure and phase changes are made in the stream databank.

**Parameters :**

stream (int) inlet stream

model (int) model for fluid flow and phase equilibria (see below)

diam (double) pipe internal diameter

rough (double) parameter defining relative pipe roughness

length (double) length of this segment

dHeight (double) height difference (inlet, outlet)

dHeat (double) heat added, removed

**Codes for models**

1 Beggs & Brill / Hazen-Williams / AGA

additional models on request to Prode

**File save**

mc\_AFSave("C:\ProgramData\prode\test.ppp") = 0    save modifications to a new archive

